

Comprehensive Two-Dimensional Gas Chromatography Coupled to Time-of-Flight Mass Spectrometry in Human Metabolomics

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1. Introduction

Metabolomics is a discipline aiming to characterize a phenotype by means of metabolome analysis. In recent years, it has developed into an accepted and valuable tool in life sciences and its use has been growing rapidly in the study of microbial, plant, and mammalian metabolomes. It has been shown to be an effective tool in characterizing cancer cells and their response to anticancer drugs (Griffiths & Chung, 2008). To assess the effects of drugs on important pathways in clinical trials of innovative therapies, metabolomic approaches might be more cost-effective than those that measure specific molecular targets (Workman et al., 2006). The derived biomarkers applied in early clinical trials are expected to help identify appropriate patients, provide proofs of concepts, aid decision making, and ultimately reduce the high level of attrition and costs of drug development (Sarker & Workman, 2007).

The biological specimens used in human metabolomic studies are e.g. urine (Weiss et al., 2007), blood plasma (Boernsen et al., 2005), and saliva (Walsh et al., 2006). All of them can play an important role in the diagnostic processes of problem illnesses. The individual metabolome of human biofluids is defined by genetic factors but it can also be affected by diet, age, disease, etc. In this respect, the use of human cell cultures offers a good alternative, since the influence of the above-mentioned factors is minimized in a culture where a defined extracellular environment takes place (Rabinowitz et al., 2006).

As concerns the analytical techniques applied in metabolomics, chromatographic techniques coupled to mass spectrometry play an important role. For the analysis of organic acids, amino acids, and sugars gas chromatography-mass spectrometry (GC-MS) after derivatization is widely applied. Since biological materials represent a very complex matrix,

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classical GC-MS techniques can struggle with the high number of components present and the occurring co-elutions. In this respect, comprehensive two-dimensional gas chromatography (GC \times GC) brings significant benefits and its coupling with time-of-flight mass spectrometry (GC \times GC-TOF-MS) has become an emerging technique in this field (Koek et al., 2011).

Due to a polar nature of target compounds, derivatization procedure is required for GC-MS analysis. Although silylation is the most widely used approach, it has certain limitations such as formation of more products from a single analyte. Moreover, the ratios between individual silylated products can change with time. To overcome these drawbacks, other derivatization procedures such as indirect alkylation via chloroformates or acylation have been used (Husek & Simek, 2006).

This contribution is focused on the application of comprehensive two-dimensional gas chromatography coupled to time-of-flight mass spectrometry for the analysis of human biological materials (urine, plasma, and cultured skin fibroblasts) in relation to diagnosing metabolic disorders, cell metabolism quenching, and drug metabolomic impact prediction.

2. Comprehensive two-dimensional gas chromatography

2.1 Basic theory

Comprehensive two-dimensional gas chromatography (GC \times GC) is a technique utilizing two columns of different selectivity connected in series by the modulation device. The modulator cuts slices from the first-dimension column effluent and re-injects them to the secondary column. Due to the difference in column polarity, each simple compound is subjected to two independent separation mechanisms. Compared to one-dimensional GC, this technique brings dramatically increased peak capacity, improved peak resolution, and up to an order of magnitude increase in compounds' detectability. In contrary to heart-cutting variety, in GC \times GC all effluent from the primary column passes through the secondary column, maximizing sample resolution throughout the entire analysis (Gorecki & Harynuk, 2004; Beans & Brinkman, 2005). A theoretical and practical comparison of one-dimensional GC and GC \times GC in terms of peak capacity has been published by Blumberg et al., 2008.

In GC \times GC, two basic orthogonality rules should be kept: (i) independence of separation mechanisms, i.e. the two columns should possess of different selectivity; (ii) preserving of the first-dimension separation, i.e. the peaks already separated on the first column must not be mixed-up in the modulator. For this reason, the modulation must occur at frequency of at least 3-5 modulations per first dimension peak. But in practice, the full independence is not possible, so in the case of GC \times GC this must be considered as the degree of GC \times GC system orthogonality, or "relative" or "partial" orthogonality which can be characterized by the percent usage of the available separation space (Ryan et al., 2005; Zhu, 2009).

2.2 Technical aspects

GC \times GC occurs by the subsequent re-injection of effluent from one chromatographic column into the second "orthogonal" column. As already mentioned, a minimum number of modulations per first dimension peak are required to maintain the first dimension separation which typically results in modulation periods of 1-5 s. A flash separation on the

second dimension column has to be completed before the next modulation cycle starts. In this way, the separation obtained in the first dimension is preserved and additional separation on the second column is obtained. The re-injection process is called modulation and is enabled by an interface device called modulator, often referred to as the “heart” of the system. A GC \times GC modulating interface can be placed at the end of the first dimension (1D) or at the beginning of the second dimension (2D), e.g. for thermal modulators, or between the columns (valve-based modulators). Nowadays, cryo-modulators which trap primary column effluent below ambient temperatures with the use of various cooling mechanisms are most commonly used (Edwards et al., 2011).

As a typical column set-up, a nonpolar column is used as the 1D and polar one in the 2D . Under these conditions, separation according to the volatility of the compounds occurs in the 1D column while “polarity” separation dominates in the 2D column. In principle, the column arrangement can be inverted. Both set-ups have their advantages and disadvantages and therefore the right set-up must be a result of optimization of a particular application. As concerns the column dimension, the first column is relatively long (typically 30-60 m) and normal bore (typically 0.25 mm with 0.25 μm film thickness). The second column has to be very short and narrow (typically 1-2 m of a 0.1 mm column with 0.1 μm film thickness) to perform a very fast separation. The use of such narrow-bore column results in limited sample capacity and easy 2D column overloading, especially for biological samples. This limitation has been overcome by the use of wider-bore columns in the second dimension (Koek et al., 2008).

Primary GC \times GC data are a series of second dimension chromatograms registered by the detector. Appropriate software reconstructs the second dimension chromatograms into three-dimensional plots or contour plots as shown in Figure 1.

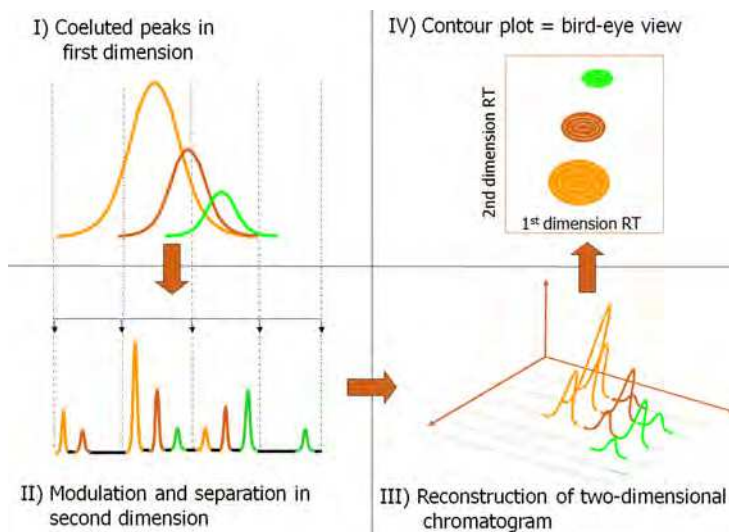


Fig. 1. Construction of a contour plot. The 1D effluent containing not fully separated analytes is modulated to create a series of short 2D chromatograms. They are reconstructed by the software form a three-dimensional view. For the practical purposes the contour plot view is more feasible.

As concerns the detectors applicable in GC \times GC, the peak width generated by this technique must be taken into account. Typical GC \times GC peaks are 0.1-0.3 s wide, thus to get a sufficient number of data points to accurately describe the shape of the peak (at least 10 points per peak) a detector must collect data at a rate of at least 100 Hz. For the coupling of GC \times GC with MS detection, a high-speed TOF-MS is the technique of choice in most studies. Quadrupole MS have been also applied in GC \times GC (Adahchour et al., 2006) with some compromises in the data density and mass range.

Since GC \times GC generates large quantities of data, appropriate software tools become very important, not only for data acquisition but also visualization and interpretation. Depending on the data itself, various operations may be required, typically including background removal, mass spectral deconvolution, peak finding, combination of modulated peaks, peak height (area, or other characteristics) computation, identification of the found peaks by the comparison with mass spectral databases (or custom libraries), and finally export of analysis report (Reichenbach et al., 2004).

2.3 Application of GC \times GC and its current trends

Since its introduction in 1991 (Liu & Philips, 1991), the GC \times GC technique has gone through the years of rapid development. Today, GC \times GC is widely used in many diverse areas which cover an interesting variety of applications. In general, the usage possibilities of GC \times GC can be divided into three areas – fingerprints of very complex matrices, target analyses, and identification of unknown compounds. An excellent review on the GC \times GC applications written by Adahchour et al. maps the usage of the technique from its introduction till 2008 in following fields – petrochemical products, environmental studies (soils and sediments, airs and aerosols, cigarette smoke), organohalogen compounds, food analysis (fats and oils, essential oils, alcoholic beverages), and also biological samples (Adahchour et al., 2008).

A significant progress can be noted in biosciences applications. In the field of human metabolomics, it covers e.g. metabolomic profiling of infant urine (Kouremenos et al., 2010; Wojtowicz et al., 2010), biomarker discovery of diabetes mellitus (Li et al., 2009), defining the “metabolome” of psychological disorders like schizophrenia (Oresic et al., 2011), sterol analysis (Mitreviski et al., 2008), analyses of tumorigenic cells (Paskanti et al., 2010), identification of anabolic agents in doping control (Mitreviski et al., 2010), or enantioselective analyses (Wadhier et al., 2011).

As regards the recent developments in GC \times GC, the utilization of new stationary phases such as ionic liquids (Zapadlo et al., 2011), development of new modulators, e.g. (Panic et al., 2011), and improvement of data handling and evaluation (Wang et al., 2010; Kim et al., 2011; Koek et al., 2011) should be mentioned.

3. Derivatization via chloroformates

Analytes derivatization is employed in many analytical methods that utilize GC as a final step. The dominant reasons for derivatization in GC are either to increase analytes volatility, to improve their chromatographic behaviour by decreasing of polarity or to increase the detector sensitivity of the target analytes. The group of silylation procedures is by far the dominant derivatization methods. Silylation is almost universal technique and the silyl groups increase the total ion current which leads to increase the sensitivity using positive ion MS (Wells, 1999).

On the other hand, this derivatization procedure has several disadvantages. First, it is time-consuming and requires elevated temperature. Further, the silylation reactions result in the formation of many artifacts as well as forming of multiple peaks for the same compound or the presence of unexpected peaks in the chromatogram. Also, in the electron impact mass spectra of silylated compounds, the non-specific masses belonging to the silyl group prevail, while the molecular and other characteristic ions have low intensity, which complicates spectra interpretation. Finally, it is also important to mention the instability of the silylated compounds and its sensitivity even to the traces of moisture which makes sample preparation and storage more demanding (Little, 1999; Ong et al., 2010).

As a promising alternative derivatization technique, an indirect alkylation via chloroformates appears. The sample preparation procedure is as follows. A portion of pyridine (serving as a catalyst) and an alcohol (to form esters) is added to the sample that is present in basic-aqueous environment. The reaction itself starts with the addition of an appropriate alkyl chloroformate. The reaction is fast (seconds) and no heating is needed. An illustration of derivatization reaction is shown in Figure 2. After the reaction, the derivatives are directly extracted into the water-immiscible organic solvent (chloroform, isoctane) that can be immediately (or after drying by e.g. anhydrous sodium sulfate) injected. In this way a biological material (plasma, urine, cell extract) can be derivatized without any pretreatment.

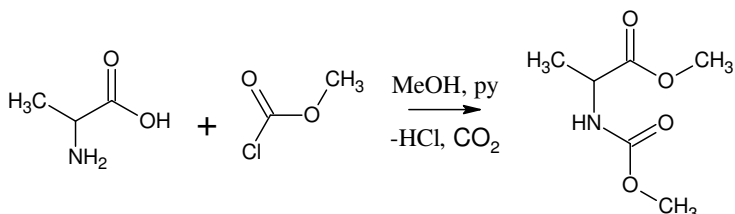


Fig. 2. Derivatization method using methyl chloroformate (MCF). 2-Aminopropanoic acid (alanine) is converted to methyl 2-[(methoxycarbonyl)amino]propanoate

Using this procedure, due to the reaction conditions (pH, presence of other solvents), many polar-functional groups are converted to the corresponding forms, i.e. - carboxy, amino, hydroxyl, and thiol groups to esters, carbamates, carbonates, and thiocarbonates, respectively. The reactions are robust with low-cost reagents and in majority cases (>95%) produce a single stable derivate that has an easily interpretable MS spectrum (Figure 3).

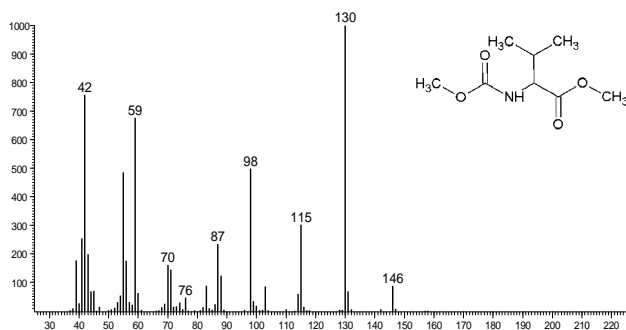


Fig. 3. MS spectrum of MCF derivate of valine

4. Material and methods

4.1 Chemicals and reagents

Internal standards (4-phenylbutyric acid for TMS, norvaline for MCF), ethoxyamine hydrochloride, methanol (HPLC grade), pyridine (p.a.), methyl-chloroformate (MCF, 99%, for GC), chloroform ($\geq 99.9\%$, for HPLC), trypsin/EDTA (10 \times), N-methyl-N-trimethylsilyltrifluoroacetamide (MSTFA), containing 1% of trimethylchlorosilane, Dulbecco's Modified Eagle's Medium, and amphotericin were purchased from Sigma-Aldrich (St. Louis, USA). Fetal bovine serum was from PANBiotech (Aidenbach, Germany), sodium chloride solution (0.9%) from B Braun (Melsungen, Germany). Other chemicals for sample preparation, i.e. hydrochloric acid, sodium chloride, sodium hydroxide, anhydrous sodium sulfate, sodium bicarbonate, ethyl acetate (p.a.), and acetone (p.a.) were supplied by LACH-NER (Neratovice, Czech Republic). 5-Fluorouracil (250 mg in 5 mL) was from EBEWE Pharma (Unterach, Austria). Standard mixture of 32 amino acids and dipeptides (200 $\mu\text{mol/L}$) was from the EZfaast kit (Phenomenex, USA). All chemicals and reagents were of analytical grade or higher.

4.2 Samples and their preparation

4.2.1 Urine samples

For the analysis of organic acids, urine samples were acidic-extracted into ethyl acetate, ethoxymated and silylated by MSTFA. We analyzed 10 healthy urines and urines from patients with inherited metabolic disorder. For more detail see (Wojtowicz et al., 2010).

Analytes response was normalized to the concentration of creatinine measured by the common Jaffe rate method.

4.2.2 Fibroblasts

Human skin fibroblasts were cultured by standard protocol (Dulbecco's Modified Eagle's Medium, supplemented with 10% of fetal bovine serum and amphotericin 100 $\mu\text{g/mL}$, 37 °C, 5% CO₂, 25 cm² flasks) to confluence. The cells were harvested by quenching or trypsinization (see below).

Quenching procedure: The cells were quenched by spraying-out of 20 mL of 60% aqueous methanol (v/v) pre-cooled to -50 °C using plastic syringe with bent needle. The flasks with quenched cells were kept on dry ice and the cells were extracted with 1 mL of cold (-50 °C) methanol solution (80%, v/v) while scraping. The cell debris in the methanol solution was drained out with pipette and another 1 mL of cold extraction solution to wash the flask was used. Both methanol fractions were combined, sonicated (1 min), and centrifuged (1800g, 5 min) to remove the cell pellet and the supernatant was freeze-dried, silylated or alkylated via MCF. For the experiment with 5-fluorouracil (5-FU), the growing medium was supplemented with the drug (50 $\mu\text{mol/L}$) for 6, 24, and 48 hours before quenching procedure, respectively.

Trypsinization: Before trypsinization the cells were washed twice with 0.9% sodium chloride solution and then the trypsin/EDTA solution was added. After 2 min, trypsin was deactivated by adding of 5 mL of a cultivation medium. The cell suspension was centrifuged

(260g, 5 min) and the pellet was extracted twice by 1 mL of 80% methanol. Combine extracts were freeze-dried.

Freeze-dried intracellular metabolite extracts were derivatized for analysis by two-stage silylation procedure based on the method described previously (Koek et al., 2006). The dry extracts were derivatized with ethoxyamine hydrochloride (10 μ L, 56 mg/mL in pyridine) and 20 μ L of pyridine for 60 min at 40 °C. Subsequently, the extracts were silylated for 50 min at 40 °C with 40 μ L of MSTFA and 30 μ L of pyridine.

The MCF derivatization was as follows: Dry metabolite extract was dissolved in 100 μ L of water and 200 μ L of sodium hydroxide (0.5 mol/L). The mixture was transferred into the glass tube containing 20 μ L of internal standard norvaline (0.1 mmol/L). After, 200 μ L of methanol and 50 μ L of pyridine as a catalyst were added and the mixture was briefly vortexed. The derivatization reaction was started by adding 20 μ L of MCF and the mixture was then vortexed for 30 s. Another 20 μ L portion of MCF was added again followed by shaking for 30 s. To separate the MCF derivatives from the reactive mixture a 300 μ L of chloroform was added and shaken 10 s followed by the addition of 300 μ L of sodium bicarbonate solution (50 mmol/L) and shaking for an additional 10 s. For better layering, the tubes were centrifuged (1000g, 1 min). The upper aqueous layer was discarded and the chloroform phase was dried by adding a small portion of anhydrous sodium sulfate. The dry organic solution was transferred to a GC vial with an insert which was tightly capped and then analyzed.

4.2.3 Plasma samples

The control and patient plasma samples were from infants from routine diagnostic processes performed in the laboratory of authors. The diagnoses had been previously confirmed by biochemical, enzyme or molecular-genetic analyses in all the patients. We analyzed 10 control samples and 19 samples with amino acids defects (phenylketonuria, PKU, maple syrup urine disease, MSUD, tyrosinemia I, TYR I, homocystinuria, HCYS, carbamoyl phosphate synthetase deficiency, CPS, ornithine transcarbamylase deficiency, OTC, and non-ketotic hyperglycinemia, NKH) (Janeckova et al., 2011).

For the MCF derivatization procedure, 50 μ L of plasma were pipetted into the glass tube containing 20 μ L of internal standard norvaline (0.1 mmol/L). After addition of 200 μ L of sodium hydroxide (0.5 mol/L), the derivatization procedure was the same as in the case of fibroblasts.

4.3 Optimized analyses conditions

A Pegasus 4D system consisting of an Agilent 7890A gas chromatograph (Agilent Technologies, Palo Alto, USA), a MPS2/CIS4/ALEX system (Gerstel, Mülheim an der Ruhr, Germany), and a Pegasus HT time-of-flight mass spectrometer (LECO Corporation, St. Joseph, USA) was used. The GC \times GC system employed a dual-stage, quad-jet modulator and a secondary oven, both built-in to the Agilent GC oven. A consumable-free option of the modulator was employed. Compressed air was used for both hot and cold modulation jets. For the hot jets the air was resistively heated, while for the cold jets the air passed through a moisture filter and was cooled by immersion cooling (-80 °C).

The nonpolar/polar (BPX5, 30 m × 0.25 mm × 0.25 μm & BPX50, 2.0 m × 0.1 mm × 0.1 μm, both Supelco) column arrangement with a modulation on the first column was chosen. The columns were connected using a SilTite Mini Union (SGE, Ringwood, Australia). The oven temperature program differs due to the derivatization procedure - TMS: primary oven temperature: 40 °C (2 min), 8 °C /min to 155 °C (0.2 min), 10 °C/min to 255 °C (0.20 min), and isocratically 300 °C (5 min); secondary oven temperature: +5 °C above the primary oven temperature; modulator temperature: +50 °C above the primary oven temperature; modulation period: 3 s (hot pulse 0.6 s), solvent delay 650 s; MCF: 60 °C (1.5 min), 15 °C/min, 300 °C (5 min), secondary oven temperature: +10 °C above the primary oven temperature; modulator temperature: +30 °C above the primary oven temperature; modulation period: 4 s (hot pulse 0.8 s), solvent delay 300 s.

Other conditions were as follows:: carrier gas: helium at the corrected constant flow 1 mL/min; splitless injection (1 min, 250 °C), 0.2-1 μL due to the application; TOF-MS: electron ionization (-70 eV); ion source temperature: 250 °C; acquired mass range: m/z 35-550; acquisition rate: 100 spectra/s; detector voltage: -1500 V; transfer line temperature: 250 °C.

ChromaTOF software v. 4.24 (LECO Corporation, USA) was used for system control, data acquisition, and data processing. The NIST/EPA/NIH Mass Spectral Library (2008) was used for tentative identification of compounds, with confirmations by retention indices comparisons. System Gerstel was controlled by Maestro software v. 1.3 (Mülheim an der Ruhr, Germany).

4.4 Data processing and quantification

An automated data processing based on the so called "Reference" was performed by the ChromaTOF software. The analyte concentration was calculated from the deconvoluted total ion current (DTIC) peak area.

Multivariate statistical data analyses (Principal Component Analysis - PCA, Hierarchical Cluster Analysis - CA) were performed using Statistica 8.0 (www.statsoft.com). A heat-map was created using the Cluster v. 3.0 software (<http://bonsai.hgc.jp>) and visualized by the TreeView v. 1.1.5 software (<http://jtreeview.sourceforge.net>).

4.4.1 Deconvoluted total ion current

Reference materials of rarely occurring pathological metabolites are not always easily available and/or their cost is considerably high. Therefore, it is a common practice to use the total ion chromatogram (TIC) signal for quantification along with the internal standard use. This approach was used also in this work. However, TIC quantification can overestimated results in case of chromatographic coelutions, which cannot be completely avoided even with GC × GC.

This obstacle was overcome in our work by using deconvoluted total ion chromatogram (DTIC) for the quantification of the analytes. Deconvolution is a mathematical algorithm that is based on the absence of spectral skew and faster acquisition rates for peak apex definition in TOF-MS data. This algorithm mathematically separates mass spectra of compounds that chromatographically co-elute. In addition to producing deconvoluted

spectra, the ChromaTOF software also allows the calculation of DTIC peak area. Deconvoluted TIC is the portion of TIC area corresponding to a particular analyte in a co-elution. An example is shown in Figure 4.

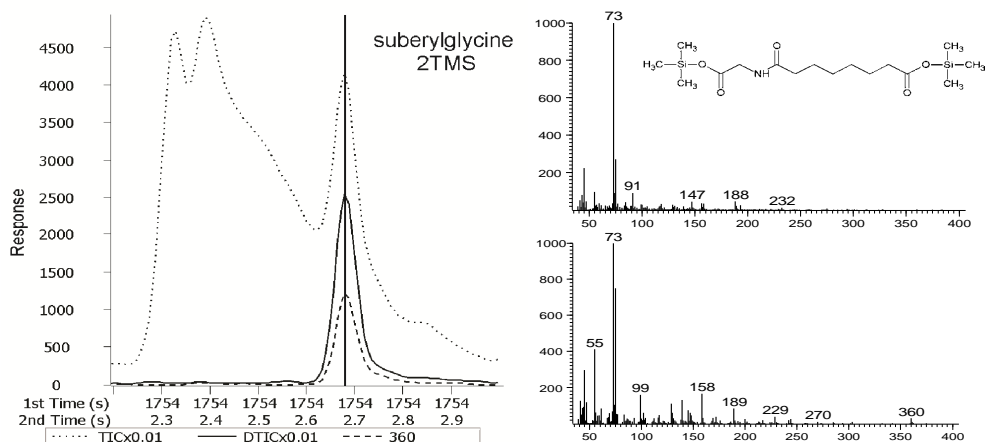


Fig. 4. Creation of DTIC for suberylglycine 2TMS (analysis of urine extract). On the right the raw peak apex spectrum (up) and the spectrum after deconvolution (down) are shown. The similarity between the deconvoluted spectra and the spectra from the library was 895.

4.4.2 Using the reference

When a classical GC-MS is used, the chromatogram is typically reviewed manually by plotting characteristic masses in the segment of the expected retention time. To avoid such a time-consuming procedure, a ChromaTOF feature called "Reference" was applied. This procedure consisted of the following steps:

- i. The sample is processed using a general peak finding method (e.g. S/N 200).
- ii. Peaks of interest are exported to the Reference, which is a set of information containing the retention times and mass spectrum of each analyte, among other data. Criteria such as the retention time window-width in both dimensions and minimum spectral match are defined by the user.
- iii. The Reference is applied to target search for each analyte in the unknown sample and for the quantification of positively identified analytes.
- iv. If some new interesting analyte is found in newly processed sample, it can be added to the existing Reference.

The retention time tolerances were determined based on the repeatability of retention times. The appropriate relative standard deviations ($n=10$) of the retention times of selected urine metabolites were under 0.17% and 1.74% for 1D and 2D , respectively (Wojtowicz et al., 2010).

Using this approach, two References have been made – one for TMS and one for MCF-based derivatization which will be described in following sections.

5. Results and discussion

5.1 Analysis of samples derivatized by silylation

EZfaast standard solution (contains 32 amino acids and dipeptides), 10 healthy and 14 pathological urine extracts, and 15 quenched fibroblast extracts derivatized by TMS were analyzed by a GC × GC-TOF-MS method and subjected to data processing procedure. After manual inspection of these peaks and confirmation of their identity by means of retention indices, a Reference was created from these compounds. This TMS compounds Reference contains 268 analytes at this moment.

The analyzed compounds are organic acids (e.g. suberate, malonate, palmitate) and their derivates (e.g. 2-oxoisovalerate, lactate, vanillylmandelate, mevalonolactone), amino acids (e.g. glycine, valine, phenylalanine) and their derivates (e.g. 3,4-dihydroxyphenylalanine, cystine, 4-hydroxyproline), N-acetylated amino acids (e.g. N-acetyl-tyrosine, N-acetyl-lysine), amines (e.g. ethanolamine, butanediamine), pyrimidines and purines (thymine, uracil, ureate), sugars (e.g. glucose), acylglycines (hippurate, propionylglycine, hexanoylglycine), and others (e.g. succinylacetone, indoleacetate, urea, cholesterol, furan-2,5-dicarboxylate).

5.1.1 Analysis of urine acidic extract in relation to organic acidurias

A large subgroup of Inherited Metabolic Disorders called organic acidurias is characteristic by increased levels of organic acids in urine or the presence of pathological ones not appearing in healthy urine. The diagnosis of organic acidurias is commonly performed by the analysis of urine after acidic extraction. Besides organic acids, some other types of metabolites also serving as pathological markers, are extracted.

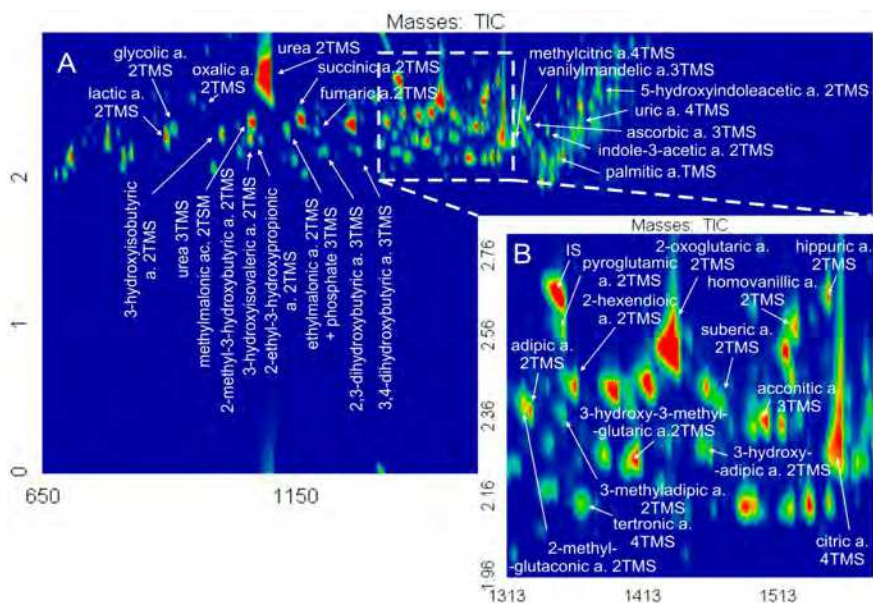


Fig. 5. GC × GC-TOF-MS contour plot from the analysis of healthy urine (A) and enlarged part (B)

Automated data processing with peak finding above S/N 200 was applied to the data, which resulted in the detection of 1353-3420 peaks in the set of studied urines. After sorting out GC column bleed peaks and the peaks belonging to the derivatization reagent, some 60% of the peaks remained. They are naturally occurring metabolites, pathological metabolites, nutrition artifacts, and drug artifacts, but most of the compounds are still not clearly identified. Our TMS Reference contains 153 identified and confirmed urine chemical species (from all 268 in the TMS Reference). In the Figure 5 is shown a 2D contour plot of healthy urine.

To evaluate the newly developed GC \times GC method as well as the data processing strategy described above, we selected external quality control samples (ERNDIM, <http://www.erndim.unibas.ch/>) and one sample from an asymptomatic patient with medium-chain acyl-CoA dehydrogenase deficiency, who had been diagnosed by neonatal screening sixth days after birth. Since the patient was in a non-crisis state, many biochemical markers of the disease were in normal levels, so he could be missed by GC-MS. Using GC \times GC-TOF-MS with the TMS Reference we found both hexanoylglycine peaks, which, although present at low concentrations, are obviously pathological markers and confirm the presence of the disease. Figure 6 illustrates the major benefit of our approach – separation of main markers of the disease from an excess of naturally occurred metabolite. For more details see (Wojtowicz et al., 2010).

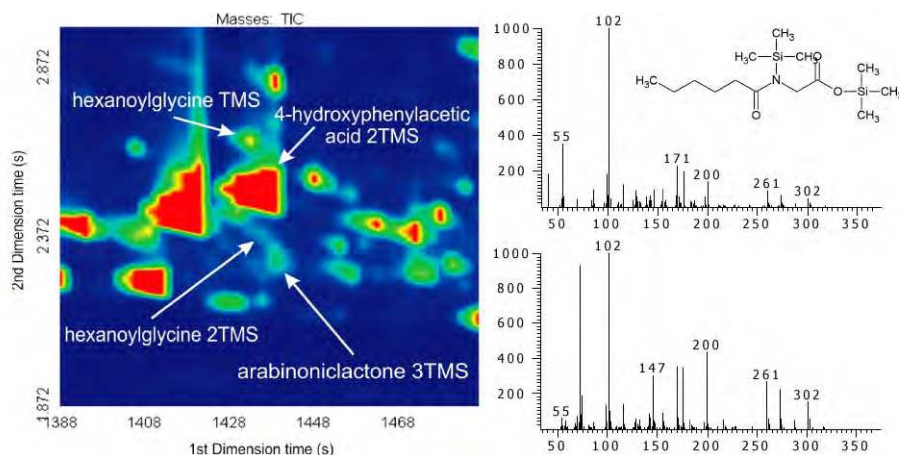


Fig. 6. Enlarged part of a contour plot from the analysis of urine from a patient with medium-chain acyl-CoA dehydrogenase deficiency. Two peaks of hexanoylglycine derivatives (1 and 2TMS) are resolved from a large peak of 4-hydroxyphenylacetic acid 2TMS. The deconvoluted spectrum (up) and the library hit belonging to the 2TMS derivate are shown (similarity 798)

5.1.2 Analysis of cultured human skin fibroblasts – comparison of trypsin treatment vs. quenching

The influence of most of external factors on the cell metabolomes is minimized in a culture where a defined extracellular environment takes place. To collect reliable metabolome data sets, culture and sampling conditions are crucial. The accurate analysis of intracellular metabolites requires a reliable sampling technique. Metabolites are generally labile species

and their reactions occur on a time scale much shorter than that of large molecule synthesis or degradation, with substantial changes in small molecule concentrations possible during a time scale of seconds (Canelas et al., 2008).

The metabolomic analysis of adherent cell cultures presents a complex challenge, mostly given by limited sample sizes. The rapid quenching of the intracellular metabolism, simultaneously with the considerable removal of superabundant growing medium, are the main prerequisites. As a result of a known rapid turnover of intracellular metabolites, the sampling process should be very fast and cause minimal metabolome loss as a result of cell leakage. In this study, we compared trypsinization as the classical procedure for adherent cell harvesting (Dettmer et al., 2011) with the simple quenching procedure developed in our laboratory.

A set of studied fibroblasts has been analyzed by a GC × GC-TOF-MS method and the data was subjected to automated data processing with peak finding above S/N 200, which resulted in the detection of 431-601 peaks. Similarly as described in the section 5.1, our TMS Reference contains 72 identified and confirmed intracellular chemical species, from all 268 in the TMS Reference (Table 1). In the Figure 7 a 2D contour plot of analysis of the intracellular metabolites extracted from spray-quenched human cultured skin fibroblasts is shown.

Acetate	2,3-Dihydroxybutyrate	4-Hydroxyproline	2-Oxovalerate
Aconitate	3,4-Dihydroxybutyrate	Cholesterol	Palmitate
Adipate	Glucose	Isoleucine	Panhotenate
Alanine	Glutamine	Itaconate	Phenylalanine
2-Aminoadipate	Glutarate	Lactate	Phosphate
2-Aminobutyrate	Glycerol	Laurate	Picolinate
4-Aminobutyrate	Glycerate	Leucine	Proline
Asparagine	Glycine	Lysine	Pyruvate
Aspartate	Fumarate	Malate	Serine
Azelaic acid	Hexenedioate	Maleate	Succinate
β-Alanine	Hippurate	Mesaconate	Stearate
Benzoate	Histidine	Methionine	Threonate
Capric acid	2-Hydroxyisobutyrate	Myo-inositol	Threonine
Citraconate	3-Hydroxyisobutyrate	Oleate	Tryptophan
Citrate	2-Hydroxybutyrate	Ornithine	Tyrosine
Cystathionine	2-Hydroxyisobutyrate	2-Oxoglutarate	Uracil
Cysteine	2-Hydroxyglutarate	2-Oxo-3-methylvalerate	Urea
Diphosphate	3-Hydroxyhippurate	5-Oxoproline	Valine

Table 1. The list of 72 unique metabolites identified and confirmed in extracts of human cultured skin fibroblasts by GC × GC-TOF-MS

In order to evaluate the effect of quenching vs. trypsinization, a cellular metabolome was determined in fibroblast cultures (n=6 for each method) from a single cell line. Data from analyses were corrected to percentages of a sum. The obtained compositional data were statistically analyzed. The geometric means for the single compounds were calculated. In order to be able to deal with Gaussian distribution of single compounds for variance analysis the data were transformed by the following equation $[1/\sqrt{2} \cdot \log(x/(1-x))]$, that represents a special case of so-called isometric log-ratio transformation (Egozcue et al., 2003). The results from trypsin and quenching approaches were compared as the natural

logarithms of ratios of means and variations (Figure 8) and multivariate statistics – PCA and CA (Figure 9). It is clearly visible that the quenching technique substantially affects the concentrations of a number of metabolites. Several metabolites (e.g. citrate, lysine) differ by an order of magnitude. From the variations it is also evident that sample preparation by means of conventional trypsinization provides substantially more variable data in comparison to quenching by means of spraying.

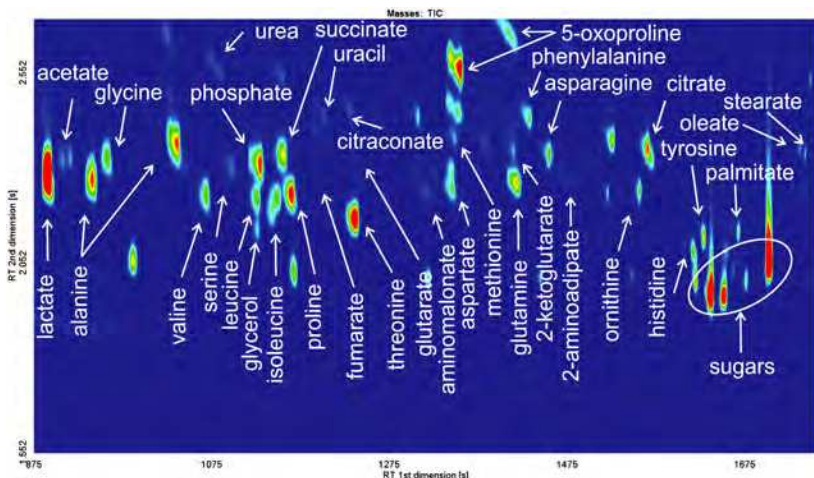


Fig. 7. GC × GC contour plot of analysis of the intracellular metabolites extracted from spray-quenched human cultured skin fibroblasts

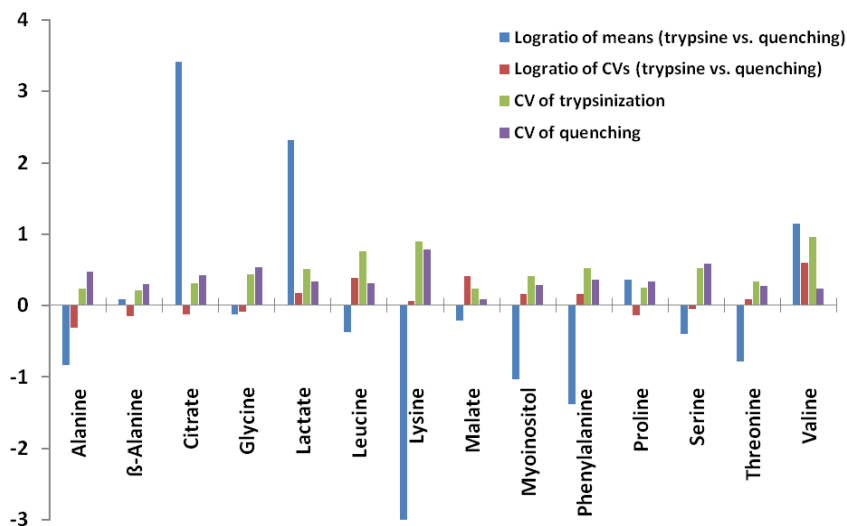


Fig. 8. The comparison of the intracellular levels of mostly deviating metabolites in trypsin treated and quenched cells. Natural logarithms of ratios of means and coefficients of variation (CV) are shown

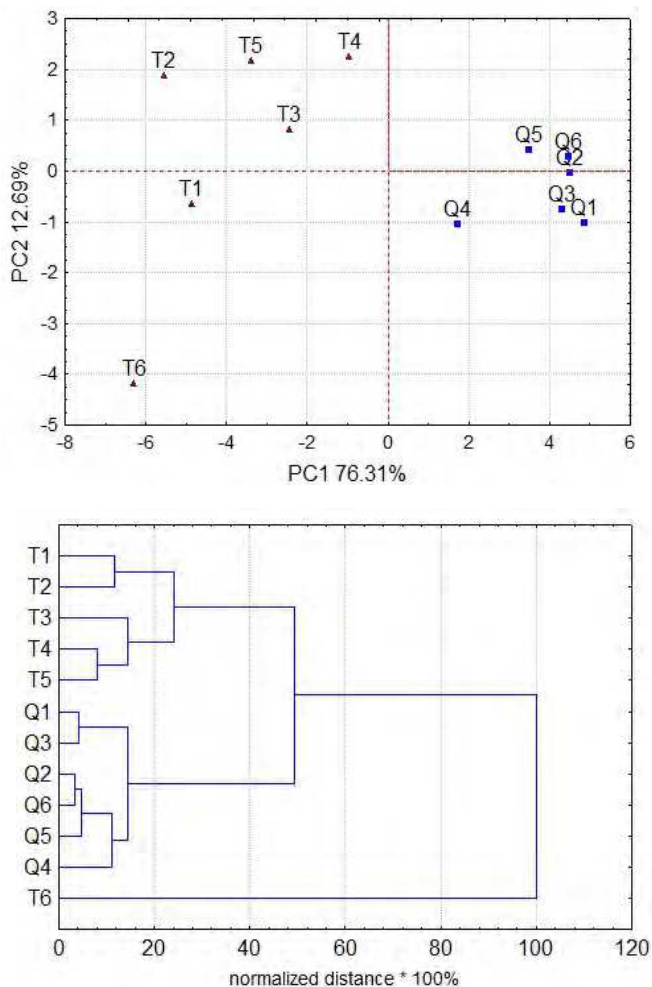


Fig. 9. PCA projection of the cases and CA dendrogram (complete-linkage, Euclidian distance) for the comparison of two cell harvesting methods (based on the listed 72 metabolites). T1-T6 trypsinization, Q1-Q6 quenching

5.2 MCF approach

EZfaast standard solution (contains 32 amino acids and dipeptides), 20 intracellular extracts from quenched fibroblasts, and 20 plasma samples derivatized by MCF were analyzed by a GC × GC-TOF-MS method and subjected to data processing procedure similarly as described in previous sections and the MCF Reference has been made.

Our MCF Reference currently contains 185 analytes – they are mostly organic acids and their derivates (the same as in the TMS approach) and compounds containing amino group (the exception is arginine because of the thermal instability of its MCF derivative that carries

a free guanidine group). Sugars are another group of compounds that cannot be detected by this approach. But that can be an advantage while e.g. cell extracts usually contains an excess of sugars that can make the obtained chromatograms not easily interpretable and some important markers can be masked. Table 2 shows some MCF-Reference based data (2D retention characteristics, unique masses – the characteristic mass identified by the ChromaTOF software, and three main ions from the MS spectra).

Amino acid	1D; 2D RT [s]	RI	Unique mass	Derivate Mr	Base peak	P2/%	P3/%
Alanine	476; 2.44	1142	102	161	102	42/65	59/63
Glycine	484; 2.50	1152	71	147	88	44/62	56/55
Sarkosine	504; 2.39	1179	102	161	102	42/78	59/49
Valine	556; 2.38	1255	130	189	130	42/88	59/82
2-Aminobutyric acid	564; 2.40	1268	88	175	88	44/51	56/41
Leucine	600; 2.38	1327	88	203	88	59/46	43/43
Threonine	612; 2.58	1348	115	191	115	59/56	42/37
Proline	632; 2.64	1383	128	187	41	128/82	41/37
Asparagine	636; 2.64	1390	127	204	42	127/75	56/72
Asparic acid	668; 2.56	1450	160	219	42	59/88	160/69
Glutamic acid	732; 2.54	1578	114	233	114	59/76	42/70
Methionine	740; 2.64	1596	61	221	61	59/48	115/40
4-Hydroxyproline	744; 2.71	1604	144	203	144	41/56	59/47
2-Amino adipic acid	784; 2.51	1693	114	247	114	55/70	59/69
Cysteine	784; 2.69	1694	59	193	59	42/77	44/59
Phenylalanine	796; 2.68	1720	42	237	42	91/65	59/57
Glutamine	840; 3.00	1872	84	218	84	44/42	59/41
Ornithine	884; 2.71	1937	128	262	128	42/51	59/51
Lysine	920; 2.76	2033	142	218	142	59/82	44/55
Histidine	940; 3.00	2088	59	227	59	81/76	42/73
Tyrosine	980; 3.00	2203	121	252	121	59/90	42/73
Tryptophan	1052; 3.78	2424	130	276	130	77/18	51/11

Table 2. Table of 22 selected amino acids derivatized via MCF presented in our Reference. 1D and 2D RT – retention time in first and second dimension, RI – retention index (calculated on the basis of absolute RT), P2 and P3/% – second and third most abundant peak/percentage of its intensity to the base peak

5.2.1 Analysis of cultured human skin fibroblasts – effect of cultivation with 5-FU

Since its synthesis, 5-FU has become one of the most widely used anticancer drugs for a variety of common malignancies, including cancers of the colon, breast, skin, and head and neck. 5-FU has been used as a component of both first-line chemotherapy regimens and in salvage regimens. Despite extensive clinical experience with 5-FU and its effective antitumor activity, many concerns remain about the optimal use of this agent.

5-FU is a prodrug, which is subject to both anabolism and catabolism. The cytotoxic activity of 5-FU depends on its anabolism to nucleotides, which exert their effects through inhibition

of thymidylate synthase activity or incorporation into RNA and/or DNA. The catabolism of 5-FU has been better understood only in recent years. The products of 5-FU catabolism have been linked to several 5-FU toxicities, including neurotoxicity (Grem, 2000; Kuhn, 2001).

This study was to show the differences in the fibroblasts metabolome after cultivation with addition of 5-FU against the non-treated controls (n=3 for each 6, 24, and 48h treatment, and without treatment, respectively), all from a single cell line. Data processing with peak

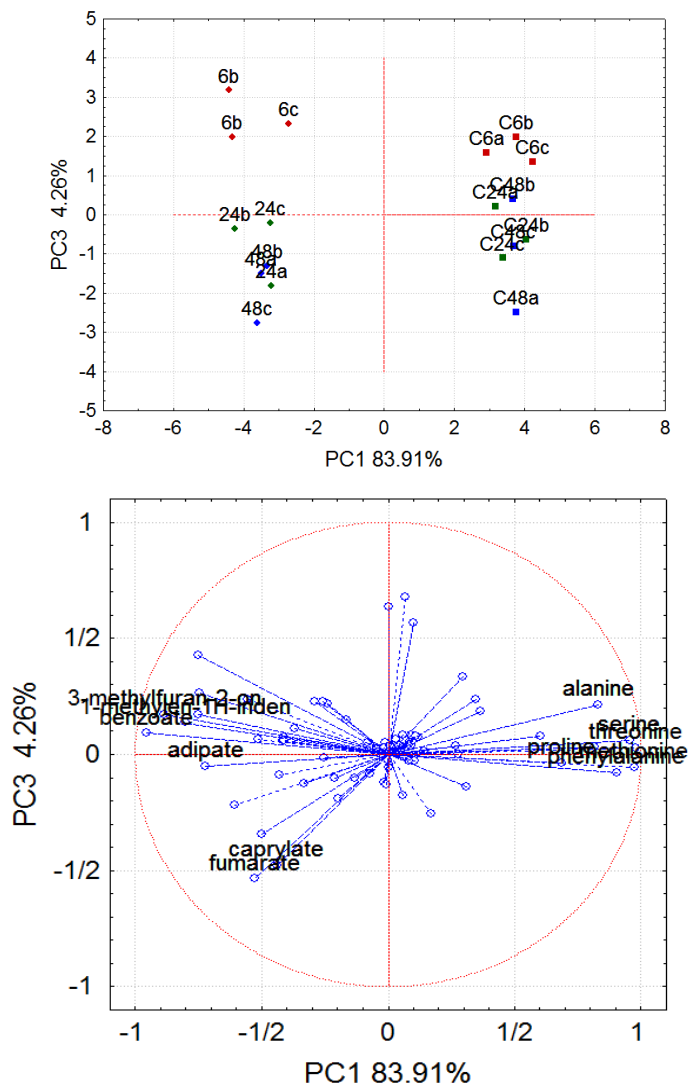


Fig. 10. PCA projection of the cases (up) and the loading plot (down) for the comparison of the effect of 5-FU on the fibroblast cultivation. Triplicates (a, b, c) for 6, 24 or 48 hours of cultivation with presence of 5-FU and controls (C) are shown

finding above S/N 200 was applied to the data, which resulted in the detection of 393-451 peaks in the set of studied fibroblasts. Our MCF Reference contains 78 identified and confirmed intracellular chemical species (from all 185 in the MCF Reference). Data from analyses were corrected to percentages of a sum, normalized to the unit standard deviation, and statistically analyzed through PCA (Figure 10). Figure 11 shows the differences for the most deviating metabolites. From the presented graphs it is clearly visible that cultivation with 5-FU influenced fibroblasts' metabolism.

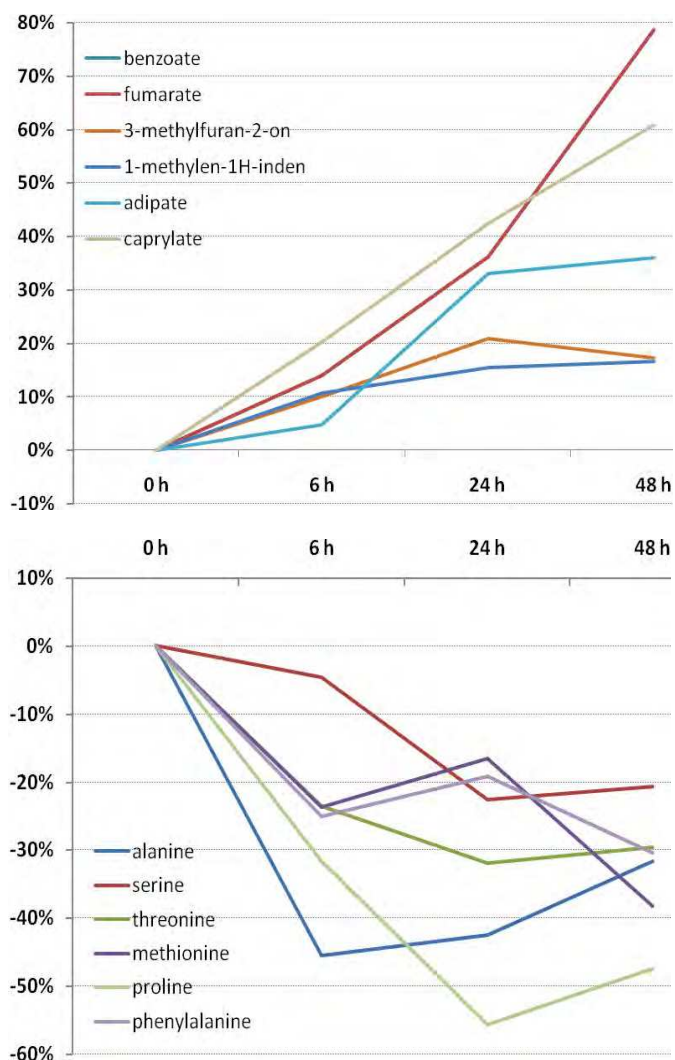


Fig. 11. Graph of the most increasing (up) and decreasing (down) intracellular metabolites influenced by 5-FU. Differences are quantified as percentages of influenced response to the non-influenced ones.

5.2.2 Analysis of human plasma in relation to metabolic disorders

In this work we focused on the diagnosis of Inherited Metabolic Disorders in plasma samples using a targeted metabolomic approach by GC × GC-TOF-MS.

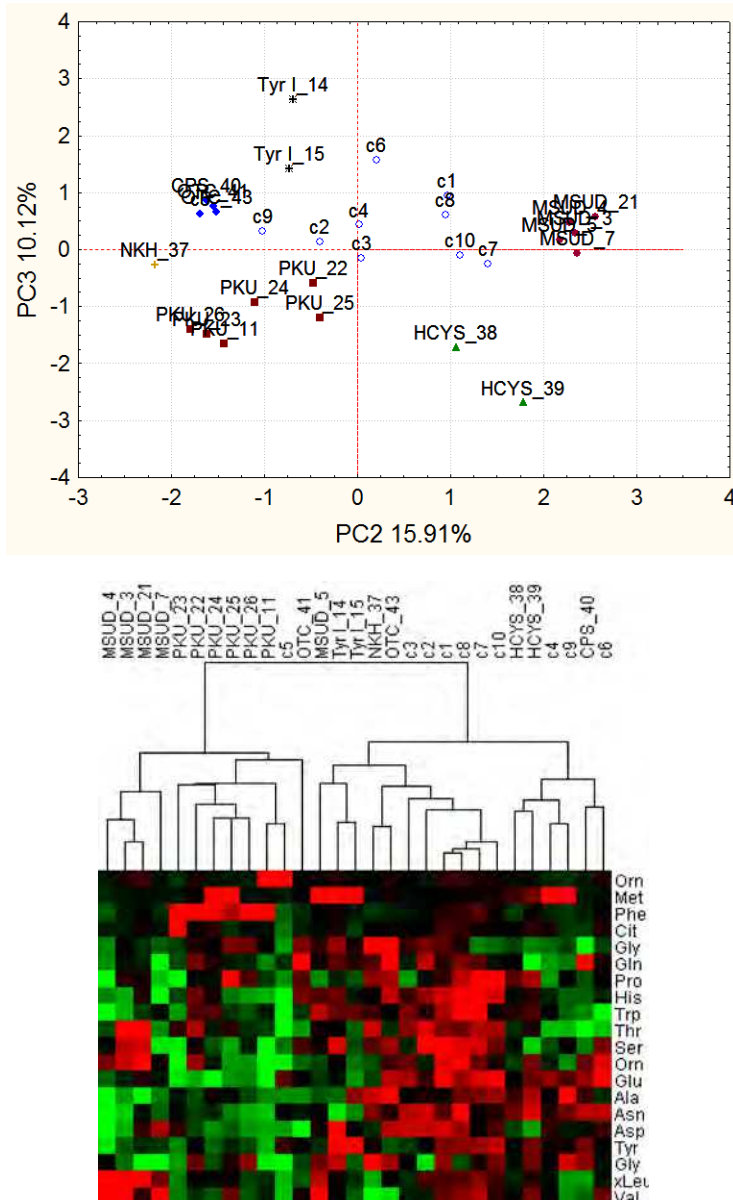


Fig. 12. PCA projection of the cases (up) and a heat map (down) visualized for selected amino acids from the analysis of human plasma

Automated data processing with peak finding above S/N 50 was applied to the data, which resulted in the detection of 408-594 peaks in the set of studied plasma samples. Our MCF Reference contains 65 identified and confirmed plasma chemical species (from all 185 in the MCF Reference).

Prior to statistical analysis the centred logratio (clr) transformation was applied. Data were evaluated using PCA and CA based on hierarchical clustering with a complete-linkage Euclidian distance method (visualized as a heat map) - Figure 12.

All the patients' samples were discriminated from the controls by appropriate metabolites in the PCA analysis. Patients with identical disease were recognized using the PCA approach and also clustered together.

6. Conclusion

This technique has been shown to be very powerful for the purpose of comprehensive sample profiling. On selected samples we demonstrated that higher separation power of GC × GC can help in removing co-elutions occurring in one dimensional approach. GC × GC is a valuable tool in metabolomic analysis of many biological matrices and enables diagnosing metabolic disorders. The great benefit is data processing that can be fully automated, what strongly simplify operator's effort and increases the sample throughput.

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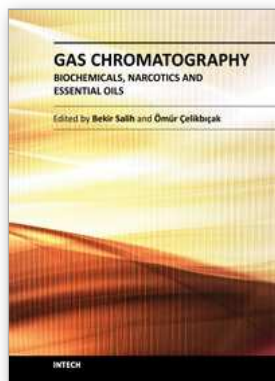
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Gas Chromatography - Biochemicals, Narcotics and Essential Oils

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Gas Chromatography involves the study of various vaporizable molecules in chemistry and the other related research fields. This analytical method has a number of features and advantages that make it an extremely valuable tool for the identification, quantification and structural elucidation of organic molecules. This book provides detailed gas chromatography information to applications of biochemicals, narcotics and essential oils. The details of the applications were briefly handled by the authors to increase their comprehensibility and feasibility. This guide should be certainly valuable to the novice, as well as to the experienced gas chromatography user who may not have the enough experience about the specific applications covered in this book. We believe this book will prove useful in most laboratories where modern gas chromatography is practiced.

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